

# **TABELLA** delle **MASSE MOLARI** (g/mol) di elementi e composti più comuni

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TABELLA 3 - *Peso formula di sostanze comuni*

<b>Ag</b>		<b>BaCO<sub>3</sub></b>	197,348
Ag	107,868	BaCl <sub>2</sub>	208,246
AgBr	187,772	BaCrO <sub>4</sub>	253,332
AgCN	133,885	BaF <sub>2</sub>	175,336
Ag <sub>2</sub> CO <sub>3</sub>	275,744	Ba(NO <sub>3</sub> ) <sub>2</sub>	261,347
AgCl	143,321	BaO	153,339
AgI	234,772	BaO <sub>2</sub>	169,338
AgNO <sub>3</sub>	169,871	Ba(OH) <sub>2</sub>	171,354
AgNCS	165,945	BaSO <sub>3</sub>	217,397
Ag <sub>2</sub> O	231,735	BaSO <sub>4</sub>	233,396
Ag <sub>2</sub> S	247,796	<b>Be</b>	
Ag <sub>2</sub> SO <sub>4</sub>	311,792	Be	9,012
Ag <sub>2</sub> CrO <sub>4</sub>	331,728	BeCO <sub>3</sub>	69,020
Ag <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub>	431,721	BeO	25,011
<b>Al</b>		<b>Bi</b>	
Al	26,981	Bi	208,980
AlCl <sub>3</sub>	133,340	Bi <sub>2</sub> O <sub>3</sub>	465,957
AlF <sub>3</sub>	83,976	BiOCl	260,432
Al <sub>2</sub> O <sub>3</sub>	101,959	Bi <sub>2</sub> S <sub>3</sub>	514,140
Al(OH) <sub>3</sub>	78,002	<b>Br</b>	
Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	342,130	Br	79,904
<b>As</b>		Br <sub>2</sub> O <sub>3</sub>	207,808
As	74,921	<b>C</b>	
As <sub>2</sub> O <sub>3</sub>	197,839	C	12,011
As <sub>2</sub> O <sub>5</sub>	229,837	CH <sub>2</sub>	14,027
As <sub>2</sub> S <sub>3</sub>	246,022	CH <sub>3</sub>	15,035
As <sub>2</sub> S <sub>5</sub>	310,142	CH <sub>4</sub>	16,043
<b>Au</b>		C <sub>6</sub> H <sub>6</sub>	78,114
Au	196,966	CN	26,017
AuCl <sub>3</sub>	303,325	CO	28,010
<b>B</b>		CO <sub>2</sub>	44,009
B	10,81	CS <sub>2</sub>	76,131
BF <sub>3</sub>	67,804	<b>Ca</b>	
B <sub>2</sub> O <sub>3</sub>	69,617	Ca	40,08
<b>Ba</b>		CaC <sub>2</sub>	64,102
Ba	137,34	CaCN <sub>2</sub>	66,097

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CaCO <sub>3</sub>	100,088	Cs	
CaCl <sub>2</sub>	110,986	Cs	132,905
CaF <sub>2</sub>	78,076	Cs <sub>2</sub> SO <sub>4</sub>	361,870
CaO	56,079	Cu	
Ca(OH) <sub>2</sub>	74,094	Cu	63,546
Ca(NO <sub>3</sub> ) <sub>2</sub>	164,087	CuCl <sub>2</sub>	134,452
CaHPO <sub>4</sub>	136,057	CuNCS	121,623
Ca <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	310,179	Cu <sub>2</sub> O	143,091
CaSO <sub>3</sub>	120,137	CuO	79,545
CaSO <sub>4</sub>	136,136	Cu <sub>2</sub> S	159,152
<b>Cd</b>		CuS	95,606
Cd	112,40	CuSO <sub>4</sub>	159,602
CdCl <sub>2</sub>	183,306	<b>F</b>	
CdO	128,399	F	18,998
CdS	144,460	<b>Fe</b>	
CdSO <sub>4</sub>	208,456	Fe	55,847
<b>Ce</b>		FeAs <sub>2</sub>	205,689
Ce	140,120	FeCl <sub>2</sub>	126,753
CeCl <sub>3</sub>	246,479	FeCl <sub>3</sub>	162,206
Ce <sub>2</sub> O <sub>3</sub>	328,237	FeCl <sub>3</sub> · 6H <sub>2</sub> O	270,296
CeO <sub>2</sub>	172,118	FeO	71,846
<b>Cl</b>		Fe <sub>3</sub> O <sub>4</sub>	231,537
Cl	35,453	Fe <sub>2</sub> O <sub>3</sub>	159,691
Cl <sub>2</sub> O	86,905	Fe(OH) <sub>3</sub>	106,868
ClO <sub>2</sub>	67,451	FeS	87,907
Cl <sub>2</sub> O <sub>7</sub>	182,899	FeS <sub>2</sub>	119,967
<b>Co</b>		FeSO <sub>4</sub>	151,903
Co	58,933	FeSO <sub>4</sub> · 7H <sub>2</sub> O	278,008
CoCl <sub>2</sub> · 6H <sub>2</sub> O	237,929	Fe <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	399,862
CoO	74,932	<b>H</b>	
CoSO <sub>4</sub>	154,989	H	1,008
CoSO <sub>4</sub> · 7H <sub>2</sub> O	281,094	H <sub>3</sub> AsO <sub>4</sub>	141,941
<b>Cr</b>		HAuCl <sub>4</sub>	339,840
Cr	51,996	HBr	80,912
Cr <sub>3</sub> O <sub>4</sub>	219,984	H <sub>3</sub> BO <sub>3</sub>	61,831
Cr <sub>2</sub> O <sub>3</sub>	151,989	HCN	27,025
		H <sub>2</sub> CO <sub>3</sub>	62,024

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(Segue Tab. 3)

H <sub>2</sub> C <sub>2</sub> O <sub>4</sub>	90,034	<b>K</b>	
HCl	36,461	K	39,102
HClO	52,460	KAl(SO <sub>4</sub> ) <sub>2</sub> · 12H <sub>2</sub> O	474,375
HClO <sub>3</sub>	84,458	KBr	119,006
HClO <sub>4</sub>	100,457	KBrO <sub>3</sub>	167,003
HF	20,006	KCN	65,119
HI	127,912	K <sub>2</sub> CO <sub>3</sub>	138,212
HIO <sub>3</sub>	175,909	K <sub>2</sub> CO <sub>3</sub> · 2H <sub>2</sub> O	174,242
HNCS	59,085	KCl	74,555
HNO <sub>2</sub>	47,012	KClO <sub>3</sub>	122,552
HNO <sub>3</sub>	63,011	K <sub>2</sub> CrO <sub>4</sub>	194,196
H <sub>2</sub> O	18,015	K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub>	294,189
2H <sub>2</sub> O	36,030	KCr(SO <sub>4</sub> ) <sub>2</sub> · 12H <sub>2</sub> O	499,390
3H <sub>2</sub> O	54,045	KF	58,100
4H <sub>2</sub> O	72,060	K <sub>3</sub> Fe(CN) <sub>6</sub>	329,259
5H <sub>2</sub> O	90,075	K <sub>4</sub> Fe(CN) <sub>6</sub>	368,361
6H <sub>2</sub> O	108,090	KHCO <sub>3</sub>	100,118
H <sub>2</sub> O <sub>2</sub>	34,014	KH <sub>2</sub> PO <sub>4</sub>	136,087
H <sub>3</sub> PO <sub>2</sub>	65,996	KI	166,006
H <sub>3</sub> PO <sub>3</sub>	81,995	KIO <sub>3</sub>	214,003
H <sub>3</sub> PO <sub>4</sub>	97,994	KMnO <sub>4</sub>	158,036
H <sub>2</sub> PtCl <sub>6</sub>	409,824	KNCS	97,179
H <sub>2</sub> S	34,076	KNO <sub>2</sub>	85,106
H <sub>2</sub> S <sub>2</sub> O <sub>3</sub>	114,133	KNO <sub>3</sub>	101,105
H <sub>2</sub> S <sub>2</sub> O <sub>4</sub>	82,073	K <sub>2</sub> O	94,203
H <sub>2</sub> SO <sub>3</sub>	98,072	KOH	56,109
H <sub>2</sub> SO <sub>4</sub>	194,128	K <sub>2</sub> HPO <sub>4</sub>	174,186
H <sub>2</sub> SiF <sub>6</sub>	144,090	K <sub>3</sub> PO <sub>4</sub>	212,275
		K <sub>2</sub> PtCl <sub>6</sub>	485,994
		K <sub>2</sub> SO <sub>4</sub>	174,260
		K <sub>2</sub> SiF <sub>6</sub>	220,280
<b>Hg</b>			
Hg	200,59		
Hg(CN) <sub>2</sub>	252,625	<b>La</b>	
Hg <sub>2</sub> Cl <sub>2</sub>	472,086	La	138,905
HgCl <sub>2</sub>	271,496	La <sub>2</sub> O <sub>3</sub>	325,808
HgO	216,589		
HgS	232,650	<b>Li</b>	
		Li	6,941
<b>I</b>		Li <sub>2</sub> CO <sub>3</sub>	73,890
I	126,904	LiCl	42,394
I <sub>2</sub> O <sub>5</sub>	333,804	Li <sub>2</sub> O	29,881

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LiNO <sub>3</sub>	68,944	(NH <sub>4</sub> ) <sub>2</sub> HPO <sub>4</sub>	132,053
Li <sub>3</sub> PO <sub>4</sub>	115,792	NH <sub>4</sub> HSO <sub>4</sub>	115,102
Li <sub>2</sub> SO <sub>4</sub>	109,938	NH <sub>4</sub> I	144,942
<b>Mg</b>		NH <sub>4</sub> MgPO <sub>4</sub> · 6H <sub>2</sub> O	245,403
Mg	24,305	NH <sub>4</sub> NO <sub>2</sub>	64,042
MgCO <sub>3</sub>	84,313	NH <sub>4</sub> NO <sub>3</sub>	80,041
MgCl <sub>2</sub>	95,211	N <sub>2</sub> O	44,012
MgCl <sub>2</sub> · 6H <sub>2</sub> O	203,301	NO	30,005
Mg(HCO <sub>3</sub> ) <sub>2</sub>	146,337	N <sub>2</sub> O <sub>3</sub>	76,010
MgNH <sub>4</sub> PO <sub>4</sub> · 6H <sub>2</sub> O	245,403	NO <sub>2</sub>	46,004
MgO	40,304	N <sub>2</sub> O <sub>4</sub>	92,008
Mg(OH) <sub>2</sub>	58,319	N <sub>2</sub> O <sub>5</sub>	108,007
Mg <sub>2</sub> P <sub>2</sub> O <sub>7</sub>	222,550	(NH <sub>4</sub> ) <sub>2</sub> S <sub>2</sub> O <sub>8</sub>	228,188
MgSO <sub>4</sub>	120,361	NH <sub>4</sub> VO <sub>3</sub>	116,976
MgSO <sub>4</sub> · 7H <sub>2</sub> O	246,466	<b>Na</b>	
<b>Mn</b>		Na	22,9898
Mn	54,938	Na <sub>2</sub> B <sub>4</sub> O <sub>7</sub>	201,212
MnCl <sub>2</sub> · 4H <sub>2</sub> O	167,271	NaBr	102,893
MnO	70,937	NaCN	49,007
Mn <sub>3</sub> O <sub>4</sub>	228,810	Na <sub>2</sub> CO <sub>3</sub>	105,987
MnO <sub>2</sub>	86,936	Na <sub>2</sub> CO <sub>3</sub> · 10H <sub>2</sub> O	286,137
MnS	86,998	NaCl	58,442
MnSO <sub>4</sub>	150,994	NaClO	74,441
MnSO <sub>4</sub> · 7H <sub>2</sub> O	277,099	NaClO <sub>3</sub>	106,439
<b>Mo</b>		NaClO <sub>4</sub>	122,438
Mo	95,94	NaF	41,988
MoO <sub>3</sub>	143,937	NaHCO <sub>3</sub>	84,005
MoS <sub>2</sub>	160,060	NaH <sub>2</sub> PO <sub>4</sub>	119,975
<b>N</b>		Na <sub>2</sub> HPO <sub>4</sub>	141,957
N	14,0067	NaHS	56,057
NCS	58,077	NaHSO <sub>3</sub>	104,054
NH <sub>3</sub>	17,030	NaHSO <sub>4</sub>	120,053
6NH <sub>3</sub>	102,184	NaI	149,894
NH <sub>4</sub> Br	97,942	NaIO <sub>3</sub>	197,891
NH <sub>4</sub> Cl	53,491	NaNCS	81,067
NH <sub>4</sub> F	37,036	NaNNO <sub>2</sub>	68,994
(NH <sub>4</sub> ) <sub>2</sub> Fe(SO <sub>4</sub> ) <sub>2</sub> · 6H <sub>2</sub> O	392,125	NaNNO <sub>3</sub>	84,993
		Na <sub>2</sub> O	61,978
		Na <sub>2</sub> O <sub>2</sub>	77,977
		NaOH	39,996
		Na <sub>3</sub> PO <sub>4</sub>	163,939

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APPENDICE

(Seguito Tab. 3)

$\text{Na}_4\text{P}_2\text{O}_7$	265,899	<b>Pd</b>	
$\text{Na}_2\text{S}$	78,039	<b>Pd</b>	106,4
$\text{Na}_2\text{S}_2\text{O}_3$	158,096	<b>Pt</b>	
$\text{Na}_2\text{SO}_3$	126,036	<b>Pt</b>	195,090
$\text{Na}_2\text{SO}_4$	142,036	<b>Rb</b>	
<b>Ni</b>		<b>Rb</b>	85,468
Ni	58,71	<b>RbCl</b>	120,920
$\text{NiCl}_2$	129,616	<b>Rb}_2\text{SO}_4</b>	266,990
$\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$	237,706	<b>Rb}_2\text{O}</b>	186,933
$\text{NiO}$	74,709	<b>S</b>	
$\text{NiS}$	90,77	<b>S</b>	32,06
$\text{NiSO}_4$	154,766	<b>SO}_2</b>	64,058
$\text{NiSO}_4 \cdot 7\text{H}_2\text{O}$	280,871	<b>SO}_3</b>	80,057
<b>O</b>		<b>Sb</b>	
<b>O</b>	15,999	<b>Sb</b>	121,75
<b>OH</b>	17,007	<b>SbCl}_2</b>	228,109
<b>3OH</b>	51,021	<b>Sb}_2\text{O}_3</b>	291,497
<b>P</b>		<b>Sb}_2\text{O}_5</b>	323,495
<b>P</b>	30,9738	<b>Sb}_2\text{S}_3</b>	339,680
<b>PBr}_3</b>	270,685	<b>Sb}_2\text{S}_5</b>	403,800
<b>PCl}_3</b>	137,332	<b>Se</b>	
<b>PCl}_5</b>	208,238	<b>Se</b>	78,96
<b>P}_2\text{O}_5</b>	141,942	<b>SeO}_2</b>	110,958
<b>P}_4\text{O}_6</b>	219,889	<b>SeO}_3</b>	126,957
<b>P}_4\text{O}_{10}</b>	283,885	<b>Si</b>	
<b>Pb</b>		<b>Si</b>	28,086
<b>Pb</b>	207,2	<b>SiF}_4</b>	104,079
<b>PbCO}_3</b>	267,208	<b>SiCl}_4</b>	169,898
<b>PbCl}_2</b>	278,106	<b>SiO}_2</b>	60,084
<b>PbCrO}_4</b>	323,192	<b>Sn</b>	
<b>Pb(NO}_3)_2</b>	331,207	<b>Sn</b>	118,69
<b>Pb}_2\text{O}_3</b>	462,397	<b>SnCl}_2</b>	189,596
<b>Pb}_3\text{O}_4</b>	685,596		
<b>PbO}_2</b>	239,198		
<b>PbS</b>	239,26		
<b>PbSO}_4</b>	303,256		

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*(Seguito Tab. 3)*

<b>SnCl<sub>4</sub></b>	260,502	<b>U</b>	
<b>SnO</b>	134,689	<b>U</b>	238,029
<b>SnO<sub>2</sub></b>	150,688	<b>UO<sub>2</sub></b>	270,027
		<b>U<sub>3</sub>O<sub>8</sub></b>	842,079
<b>Sr</b>			
<b>Sr</b>	87,62	<b>V</b>	
<b>SrCO<sub>3</sub></b>	147,628	<b>V</b>	50,941
<b>SrCl<sub>2</sub></b>	158,526	<b>V<sub>2</sub>O<sub>3</sub></b>	149,879
<b>Sr(NO<sub>3</sub>)<sub>2</sub></b>	211,627	<b>V<sub>2</sub>O<sub>5</sub></b>	181,877
<b>SrO</b>	103,619		
<b>SrSO<sub>4</sub></b>	183,676	<b>W</b>	
		<b>W</b>	183,85
<b>Ti</b>		<b>WO<sub>3</sub></b>	231,847
<b>Ti</b>	47,90	<b>WS<sub>2</sub></b>	247,97
<b>TiCl<sub>3</sub></b>	154,259		
<b>TiCl<sub>4</sub></b>	189,712	<b>Zn</b>	
<b>TiO<sub>2</sub></b>	79,898	<b>Zn</b>	65,37
		<b>ZnCO<sub>3</sub></b>	125,378
<b>Tl</b>		<b>ZnCl<sub>2</sub></b>	136,276
<b>Tl</b>	204,37	<b>ZnO</b>	81,369
<b>TlCl</b>	239,823	<b>ZnS</b>	97,43
<b>TlI</b>	331,274	<b>ZnSO<sub>4</sub></b>	161,426
<b>Tl<sub>2</sub>O<sub>3</sub></b>	456,737	<b>ZnSO<sub>4</sub> · 7H<sub>2</sub>O</b>	287,531